

## LAMPIRAN

## Lampiran 1. Hasil Determinasi Pegagan



**DIREKTORAT PENGELOLAAN KOLEKSI ILMIAH**  
 Jl. Raya Jakarta – Bogor Km. 46, Cibinong 16911, Kab. Bogor, Jawa Barat  
 Telepon 081110646760 Surel: [diti-pki@brin.go.id](mailto:diti-pki@brin.go.id)  
 Laman: [www.brin.go.id](http://www.brin.go.id)

Nomor : B-1406/IV/DI.05.07/5/2022  
 Lampiran : -  
 Perihal : Hasil Identifikasi/Determinasi Tumbuhan

20 Mei 2022

Yth.  
 Bpk./Ibu/Sdr(i). **Afra Shafa Ghalda**  
 NIM : 20180311103  
 Universitas Esa Unggul

Bersama ini kami sampaikan hasil identifikasi/determinasi tumbuhan yang Saudara kirimkan ke "Herbarium Bogoriense", Direktorat Pengelolaan Koleksi Ilmiah BRIN Cibinong, adalah sebagai berikut :

No.	No. Kol.	Jenis	Suku
1.	Tanaman Pegagan	<i>Centella asiatica</i> (L.) Urb.	Apiaceae

Demikian, semoga berguna bagi Saudara.

Plt. Direktorat Pengelolaan Koleksi Ilmiah  
 Badan Riset dan Inovasi Nasional

TT ELEKTRONIK

Dr. Ir. Hendro Wicaksono, M.Sc., Eng

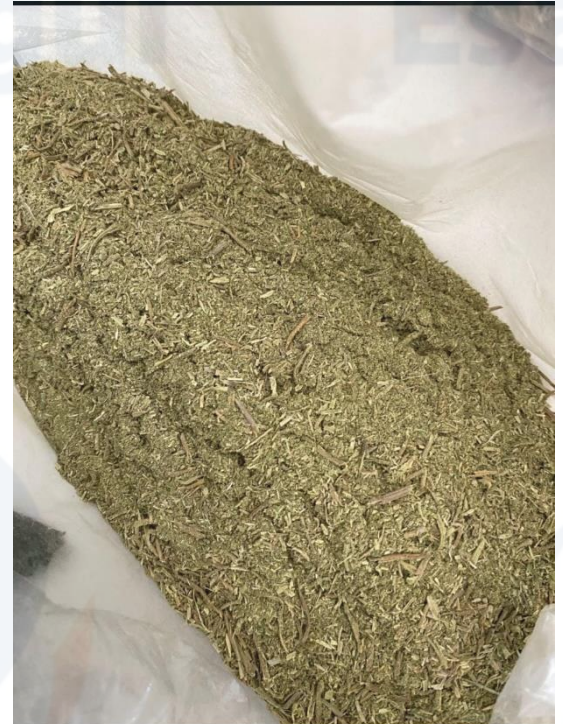


Dokumen ini ditandatangani secara elektronik menggunakan sertifikat dari BSRiE, silahkan lakukan verifikasi pada dokumen elektronik yang dapat diunduh dengan melakukan scan QR Code

Lampiran 2. Pembuatan Ekstrak Etanol 96% dan N-Heksana Daun Pegagan



Simplisia daun pegagan yang sudah kering digrinder



Diperoleh serbuk kasar



Proses maserasi ekstrak etanol 96% dan n-heksana daun pegagan



Pemisahan filtrat dan residu



Filtrat yang didapat disimpan dalam wadah yang tertutup



Proses penggantian pelarut



Filtrat dipekatan menggunakan rotary evaporator



Filtrat diupkan kembali menggunakan waterbath



Ekstrak kental etanol 96% dan n-heksana  
daun pegagan

### Lampiran 3. Skrining Fitokimia

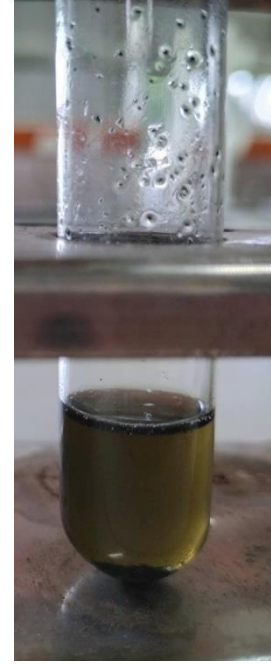
#### 1. Skrining Fitokimia Ekstrak Etanol 96%



Alkaloid (-)  
Reagen mayer



Alkaloid (-)  
Reagen dragendorff



Flavonoid (+)



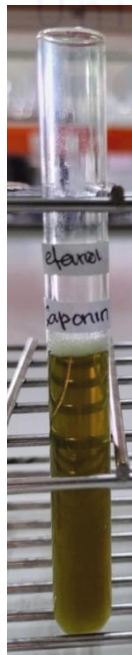
Alkaloid (-)  
Reagen Bouchard



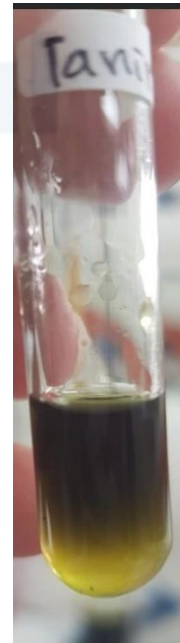
Steroid (+)



Triterpenoid (+)

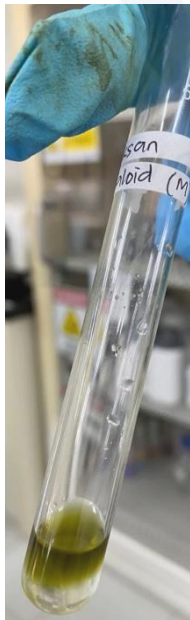


Saponin (+)



Tanin (+)

## 2. Skrining Fitokimia Ekstrak N-Heksana



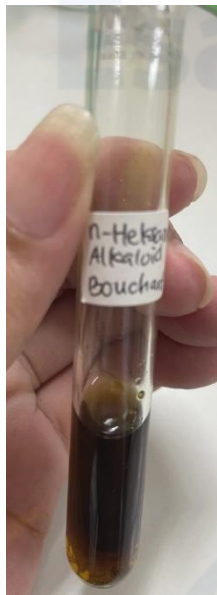
Alkaloid (-)  
Reagen mayer



Alkaloid (-)  
Reagen dragendorff



Flavonoid (+)

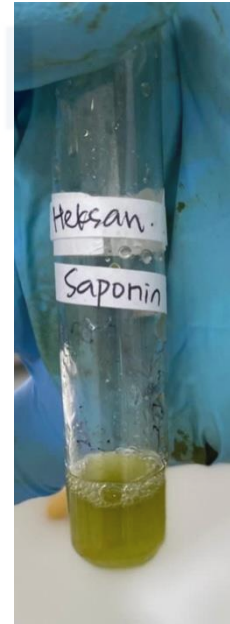


Alkaloid (-)

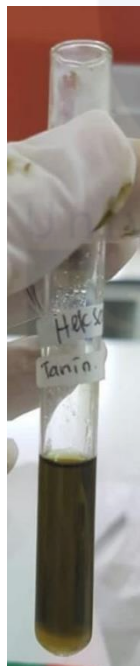
Reagen Bouchar



Steroid (+)



Saponin (+)



Tanin (+)



Triterpenoid (+)

**Lampiran 4. Hasil perhitungan Rendemen Ekstrak**

- a. Rendemen Ekstrak Etanol 96% Daun Pegagan

$$\begin{aligned}\text{Rendemen \%} &= \frac{\text{bobot ekstrak}}{\text{berat simplisia}} \times 100\% \\ &= \frac{208,62 \text{ g}}{1.250 \text{ g}} \times 100\% \\ &= 16,689\% \\ &= 16,69\%\end{aligned}$$

- b. Rendemen Ekstrak N-Heksana Daun Pegagan

$$\begin{aligned}\text{Rendemen \%} &= \frac{\text{bobot ekstrak}}{\text{berat simplisia}} \times 100\% \\ &= \frac{58,55 \text{ g}}{1250 \text{ g}} \times 100\% \\ &= 4,684\%\end{aligned}$$



**Lampiran 5. Perhitungan Larutan Sampel Uji**

## 1. Ekstrak Etanol 96% Daun Pegagan

a. 1000 ppm  
10 mg/10 mL

b. 800 ppm  
 $M_1V_1 = M_2V_2$   
 $1000 \times V_1 = 5 \times 800$   
 $V_1 = \frac{5 \times 800}{1000}$   
 $= 4 \text{ mL}$

c. 400 ppm  
 $M_1V_1 = M_2V_2$   
 $1000 \times V_1 = 5 \times 400$   
 $V_1 = \frac{5 \times 400}{1000}$   
 $= 2 \text{ mL}$

d. 200 ppm  
 $M_1V_1 = M_2V_2$   
 $1000 \times V_1 = 5 \times 200$   
 $V_1 = \frac{5 \times 200}{1000}$   
 $= 1 \text{ mL}$

e. 100 ppm  
 $M_1V_1 = M_2V_2$   
 $1000 \times V_1 = 5 \times 100$   
 $V_1 = \frac{5 \times 100}{1000}$   
 $= 0,5 \text{ mL}$

f. 50 ppm  
 $M_1V_1 = M_2V_2$   
 $1000 \times V_1 = 5 \times 50$   
 $V_1 = \frac{5 \times 50}{1000}$   
 $= 0,25 \text{ mL}$

## 2. Ekstrak N-Heksana Daun Pegagan

a. 1000 ppm

10 mg/10 mL

b. 800 ppm

$$M_1V_1 = M_2V_2$$

$$1000 \times V_1 = 5 \times 800$$

$$V_1 = \frac{5 \times 800}{1000}$$

$$= 4 \text{ mL}$$

c. 400 ppm

$$M_1V_1 = M_2V_2$$

$$1000 \times V_1 = 5 \times 400$$

$$V_1 = \frac{5 \times 400}{1000}$$

$$= 2 \text{ mL}$$

d. 200 ppm

$$M_1V_1 = M_2V_2$$

$$1000 \times V_1 = 5 \times 200$$

$$V_1 = \frac{5 \times 200}{1000}$$

$$= 1 \text{ mL}$$

e. 100 ppm

$$M_1V_1 = M_2V_2$$

$$1000 \times V_1 = 5 \times 100$$

$$V_1 = \frac{5 \times 100}{1000}$$

$$= 0,5 \text{ mL}$$

f. 50 ppm

$$M_1V_1 = M_2V_2$$

$$1000 \times V_1 = 5 \times 50$$

$$V_1 = \frac{5 \times 50}{1000}$$

$$= 0,25 \text{ mL}$$

## 3. Asiaticoside

a. 1000 ppm

5 mg/5 mL

b. 100 ppm

$$M_1V_1 = M_2V_2$$

$$1000 \times V_1 = 10 \times 100$$

$$V_1 = \frac{10 \times 100}{1000}$$

$$= 1 \text{ mL}$$

c. 50 ppm

$$M_1V_1 = M_2V_2$$

$$100 \times V_1 = 5 \times 50$$

$$V_1 = \frac{5 \times 50}{100}$$

$$= 2,5 \text{ mL}$$

d. 20 ppm

$$M_1V_1 = M_2V_2$$

$$100 \times V_1 = 5 \times 20$$

$$V_1 = \frac{5 \times 20}{100}$$

$$= 1 \text{ mL}$$

e. 10 ppm

$$M_1V_1 = M_2V_2$$

$$100 \times V_1 = 5 \times 10$$

$$V_1 = \frac{5 \times 10}{100}$$

$$= 0,5 \text{ mL}$$

f. 5 ppm

$$M_1V_1 = M_2V_2$$

$$100 \times V_1 = 5 \times 5$$

$$V_1 = \frac{5 \times 5}{100}$$

$$= 0,25 \text{ mL}$$

g. 2,5 ppm

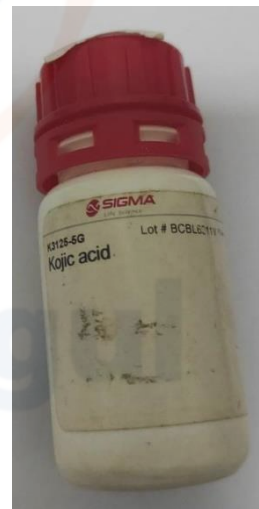
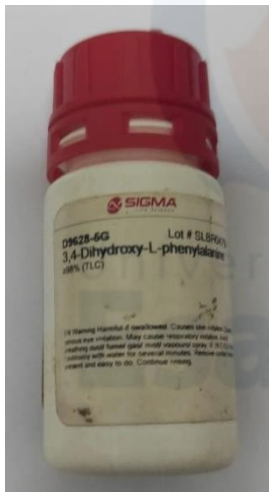
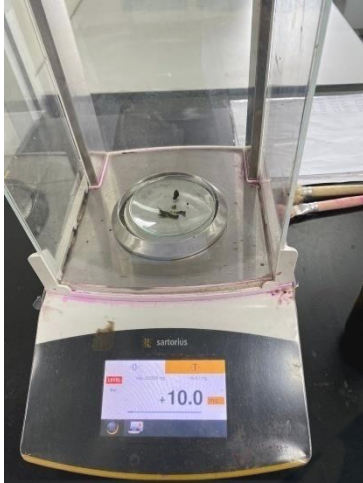
$$M_1V_1 = M_2V_2$$

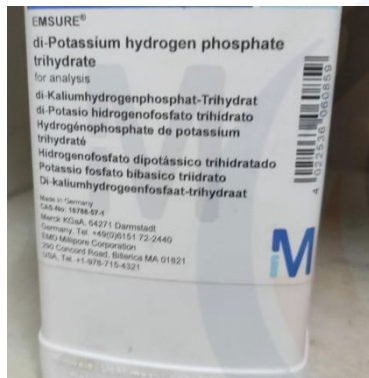
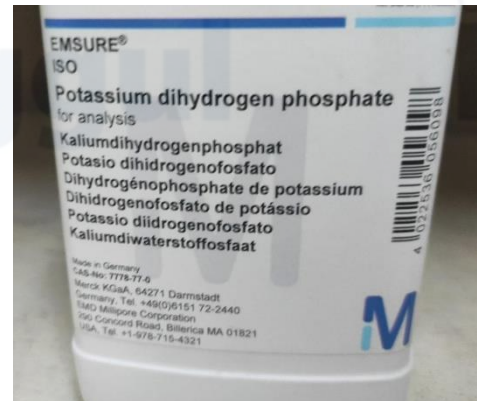
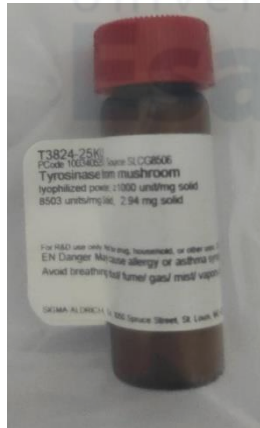
$$100 \times V_1 = 5 \times 2,5$$

$$V_1 = \frac{5 \times 2,5}{100}$$

$$= 0,125 \text{ mL}$$

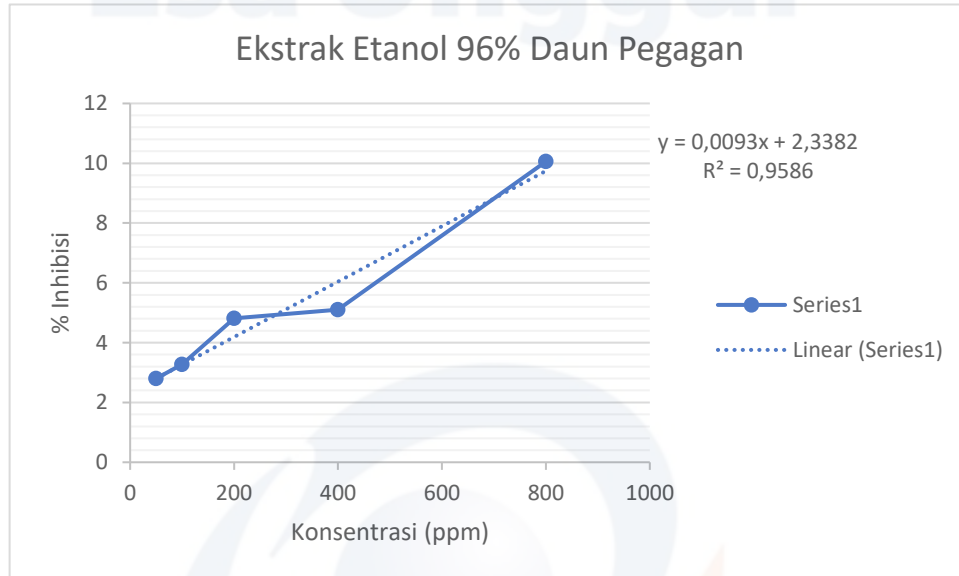
Lampiran 6. Pembuatan Larutan Sampel Uji Aktivitas Penghambat  
Tirosinase





## Lampiran 7. Grafik Uji Aktivitas Penghambat Tirosinase

### a. Ekstrak Etanol 96% Daun Pegagan



Perhitungan  $IC_{50}$  :

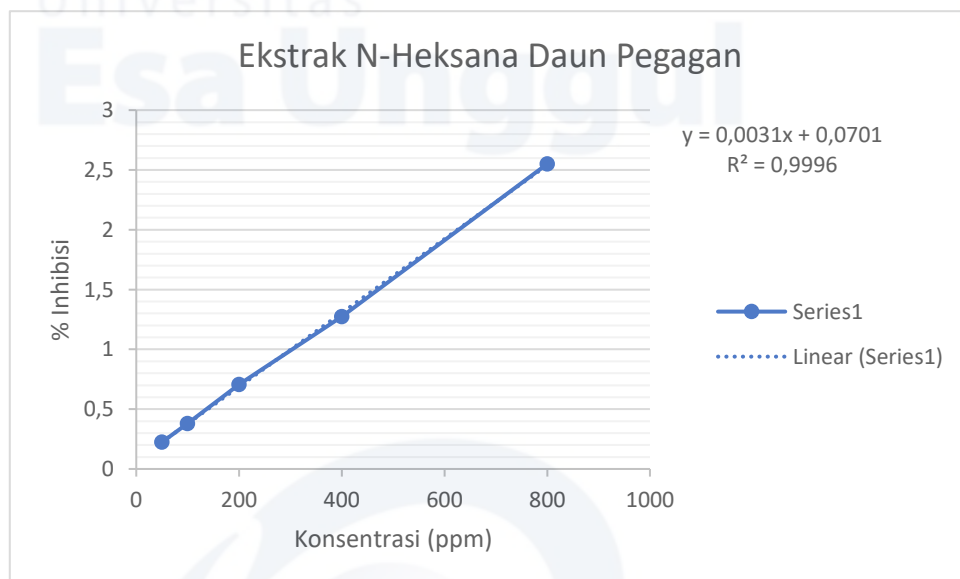
$$y = a + bx$$

$$50 = 2,3382 + 0,0093x$$

$$x = \frac{50 - 2,3382}{0,0093}$$

$$= 5149,229$$

### b. Ekstrak N-Heksana Daun Pegagan



Perhitungan  $IC_{50}$  :

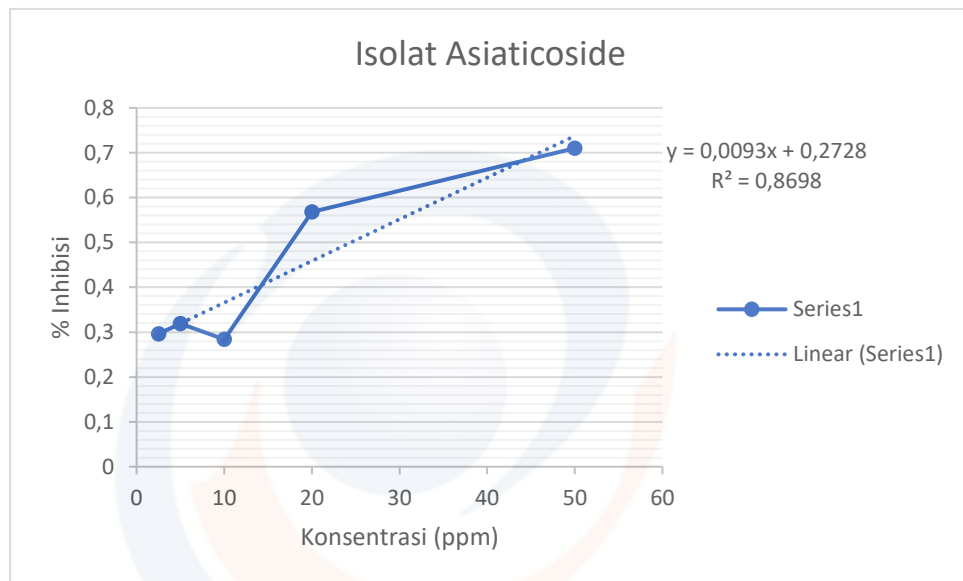
$$y = a + bx$$

$$50 = 0,0701 + 0,0031x$$

$$x = \frac{50 - 0,0701}{0,0031}$$

$$= 16171,773$$

c. Asiaticoside



Perhitungan  $IC_{50}$  :

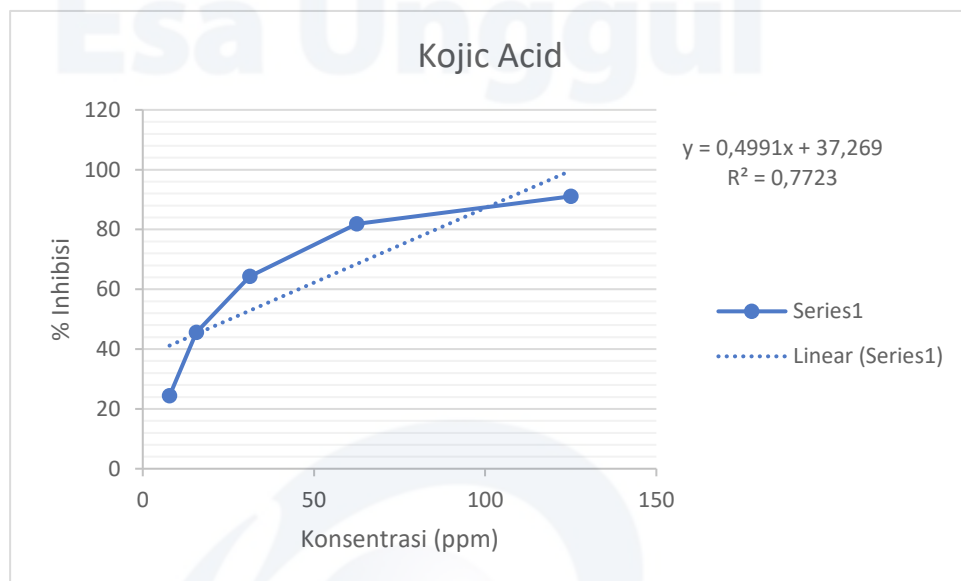
$$y = a + bx$$

$$50 = 0,2728 + 0,0093x$$

$$x = \frac{50 - 0,2728}{0,0093}$$

$$= 5351,806$$

## d. Asam kojat



Perhitungan  $IC_{50}$  :

$$y = a + bx$$

$$50 = 37,269 + 0,4991x$$

$$x = \frac{50 - 37,269}{0,4991}$$

$$= 25,507$$